

Ethyl 4-(2-hydroxyphenyl)-6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate

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Key indicators

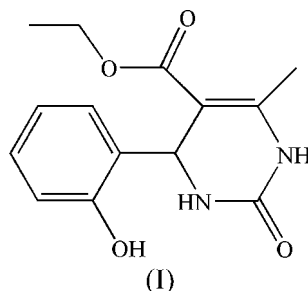
Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.048
 wR factor = 0.135
Data-to-parameter ratio = 12.8For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title molecule, $\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}_4$, all bond lengths and angles are normal. The crystal packing is stabilized by intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds involving the hydroxy and keto groups, and by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds involving the amino and keto groups.

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Comment

As part of our study of 2-oxo-1,2,3,4-tetrahydropyrimidine derivatives, which demonstrate various pharmacological activities (Kappe *et al.*, 1999; Legeay *et al.*, 2007), we report here the structure of the title compound, (I).



In (I) (Fig. 1), all bond lengths and angles agree well with those reported for related compounds (Portilla *et al.*, 2006). The hydroxy and keto groups are involved in intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1), which link the molecules into zigzag chains running along the c axis. Weak intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1) further link two neighbouring chains with the formation of eight- and 16-membered rings (Fig. 2).

Experimental

A mixture of urea (51 mmol), ethyl acetoacetate (75 mmol) and salicylaldehyde (50 mmol) in absolute ethanol (30 ml) and concentrated hydrochloric acid (10 drops) was heated to reflux for 3 h. Upon cooling to room temperature, a crude product crystallized. The precipitate was filtered off and washed with propan-2-ol. The crude product was recrystallized from propan-2-ol to afford the desired product as a colourless solid (m.p. 490 K). Colourless single crystals of (I) were obtained by slow evaporation of an aqueous ethanol (90%) solution at ambient temperature after 7 d.

Crystal data

$\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}_4$	$V = 2610.2$ (9) Å ³
$M_r = 276.29$	$Z = 8$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 20.757$ (3) Å	$\mu = 0.10$ mm ⁻¹
$b = 9.042$ (2) Å	$T = 298$ (2) K
$c = 13.919$ (2) Å	$0.32 \times 0.16 \times 0.13$ mm
$\beta = 92.313$ (3)°	

Data collection

Bruker SMART CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.967$, $T_{\max} = 0.987$
 6587 measured reflections
 2308 independent reflections
 1363 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.135$
 $S = 1.03$
 2308 reflections
 181 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.26 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.22 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1\cdots O2^i$	0.86	2.45	3.205 (5)	147
$N2-H2\cdots O3^{ii}$	0.86	2.27	3.117 (4)	167
$O4-H4\cdots O3^{iii}$	0.82	1.98	2.755 (4)	158

Symmetry codes: (i) $x, -y + 1, z + \frac{1}{2}$; (ii) $-x, -y + 2, -z + 2$; (iii) $x, -y + 2, z - \frac{1}{2}$.

All H atoms were positioned geometrically, with $C-H = 0.93-0.97 \text{ \AA}$, $N-H = 0.86 \text{ \AA}$ and $O-H = 0.82 \text{ \AA}$, and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C,N,O})$.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2000) and DIAMOND (Brandenburg, 2004); software used to prepare material for publication: SHELXTL.

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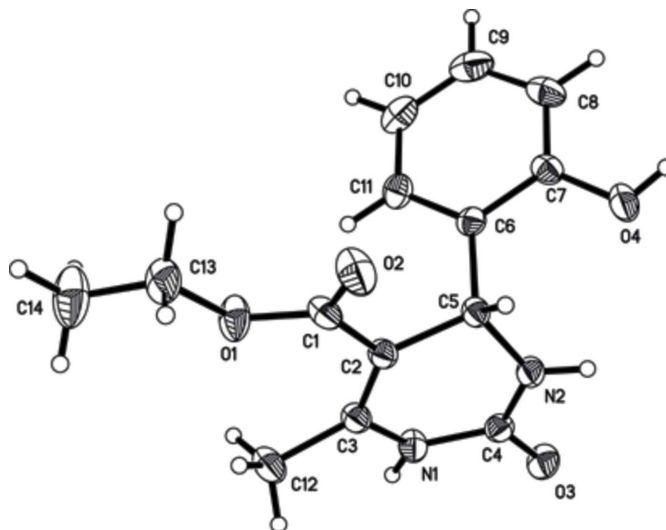


Figure 1
The molecular structure of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

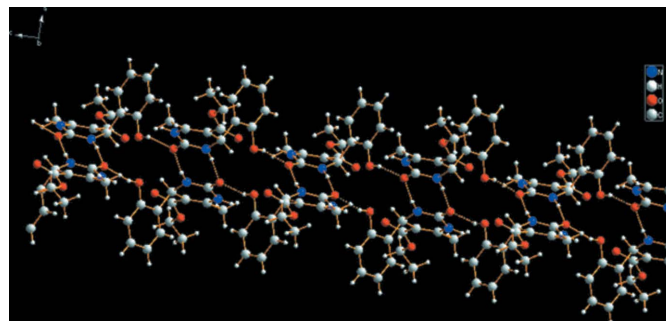


Figure 2
A portion of the crystal packing of (I), showing the intermolecular hydrogen bonds as dashed lines.

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